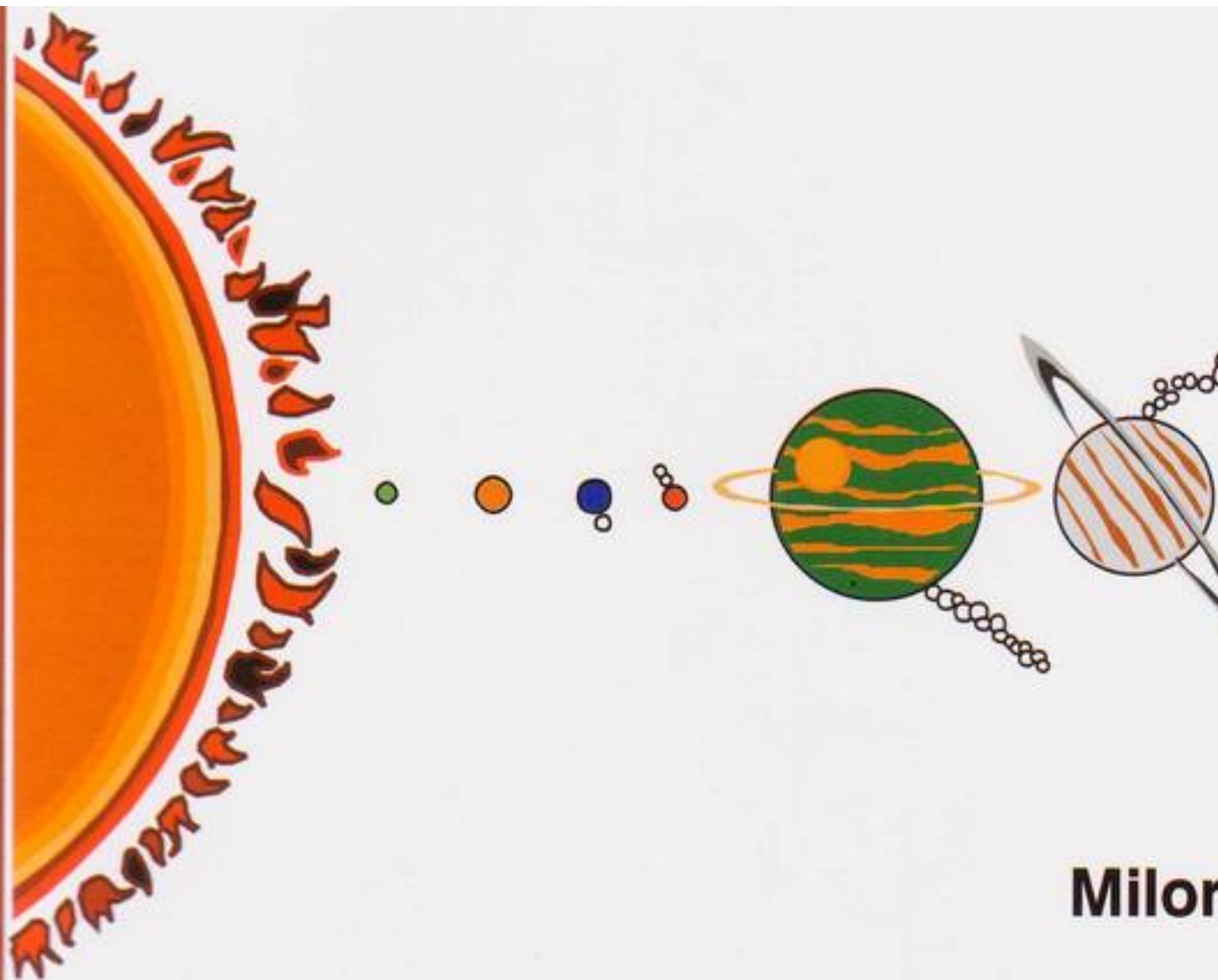
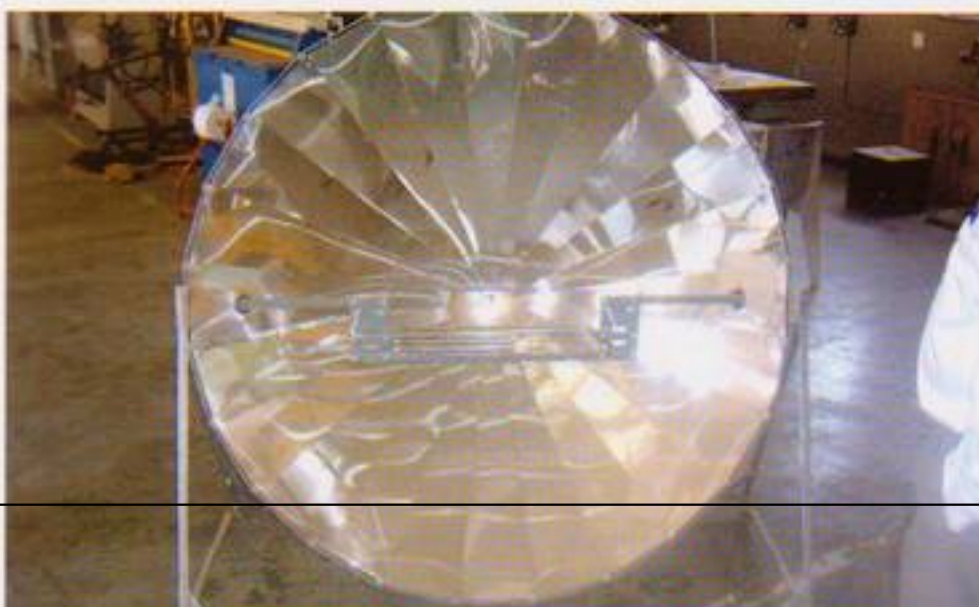




Milorad

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~~1. Zadatak: Izračunajte entalpiju stvaranja metana iz njegovih elemenata u standardnim uslovima (298 K, 1 bar). Podaci: $\Delta_f H^\circ(\text{CH}_4, \text{g}) = -74.6 \text{ kJ/mol}$, $\Delta_f H^\circ(\text{H}_2\text{O}, \text{l}) = -285.8 \text{ kJ/mol}$, $\Delta_f H^\circ(\text{CO}_2, \text{g}) = -393.5 \text{ kJ/mol}$.~~